


**SULZER**



# Propane to Propylene Technology

A low-carbon pathway for obtaining environmentally friendly propylene from propane with optional co-production of hydrogen

July 12. 2023





# Disclaimer

This presentation may contain forward-looking statements, including but not limited to, projections of financial developments, market activities or future performance of products and solutions, containing risks and uncertainties.

These forward-looking statements are subject to change based on known or unknown risks and various other factors, which could cause the actual results or performance to differ materially from the statements made herein.

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# About SULZER

We help our customers build a better world

Sulzer is a global industry leader, with two centuries of experience developing innovative products and services that drive sustainable progress — and help our customers build a better world.

3.2

Billion sales  
(CHF) 2022

12'900

Employees

40

Production  
locations

140

Service centers

100

Countries with  
Sulzer presence



# Sustainability at Sulzer

Minimize – Enable – Engage

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**Minimize**  
our carbon  
footprint

We operate in  
a sustainable way

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**Enable**  
A low carbon  
society

We contribute to  
a circular economy

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**Engage**  
our employees  
and communities

We make life better  
for those around us



# Gas to X Technologies

From 2004 to today



## 2004 - 2013

- Start of working on Gas-to-Fuel process using bromine activation
- Integrated pilot unit testing in SWRI
- Demo plant constructed & operated
- 1<sup>st</sup> commercial GTF process design

## 2014 - 2019

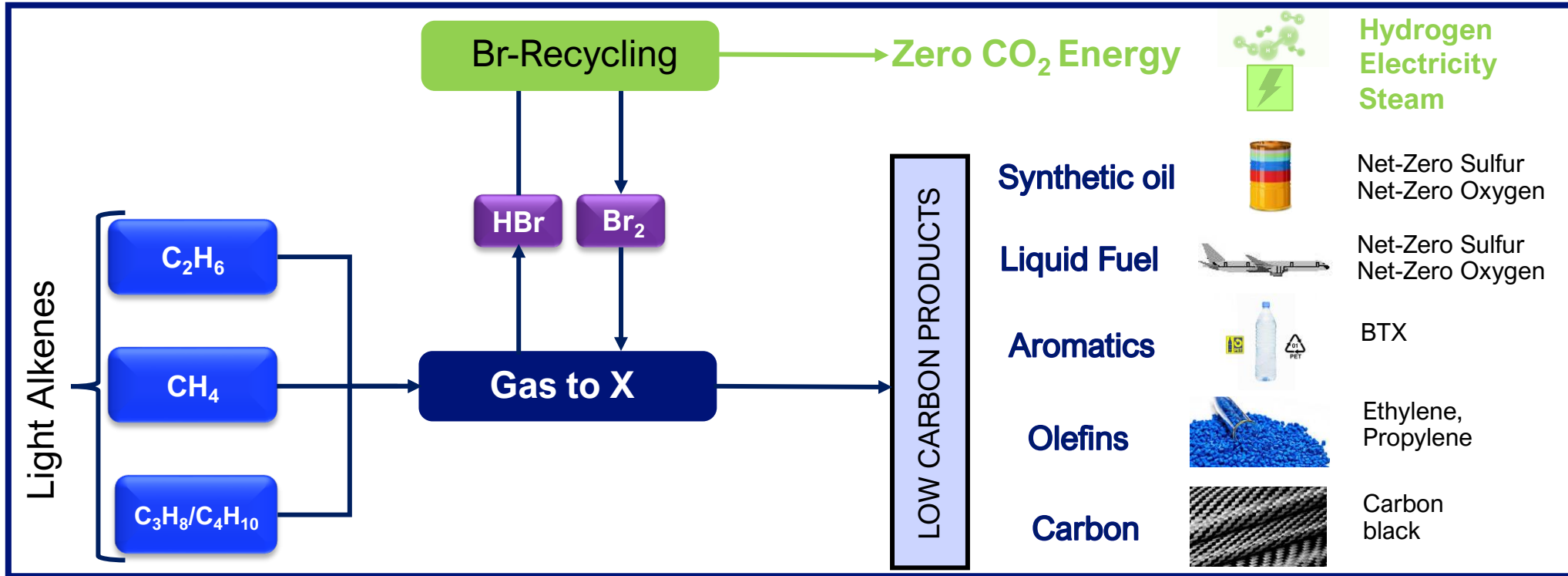
- Acquisition of the technology by GTC
- JDA with Total Energies in Methane-to-Olefins

## 2019 to today

- Acquisition of GTC by Sulzer
- Development of HBr electrolysis
- Extension of technology to C<sub>2</sub>-C<sub>4</sub> feedstock
- G2X portfolio commercialization



# Gas to X in a nutshell



> Gas to X – a universal technology portfolio for the monetization of light alkanes



# Gas to X - technology portfolio

Value adding with side products and minimizing carbon footprint

## Methane to Carbon

Decarbonization of methane to produce high value carbon with either energy self-sufficiency or energy efficient co-production of hydrogen with no impact on regional water security



## Gas to Liquid

A low-carbon, economical pathway for converting gas into a liquid to be used as synthetic oil, liquid fuel or aromatics, eliminating the need for gas flaring



## Propane to Propylene

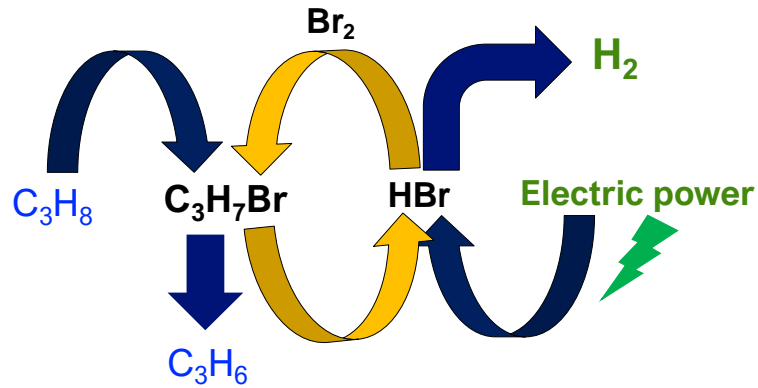
A low-carbon pathway for the production of propylene from propane that is attractive for all feed volumes but especially for small and medium feedstock sources



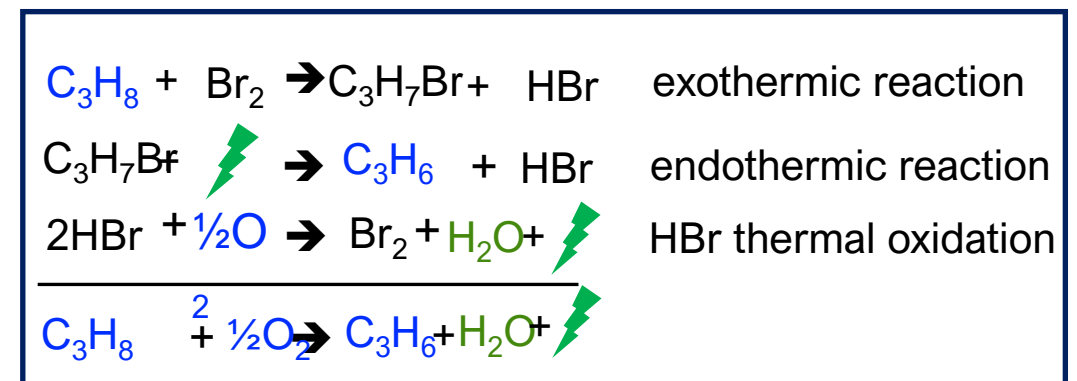
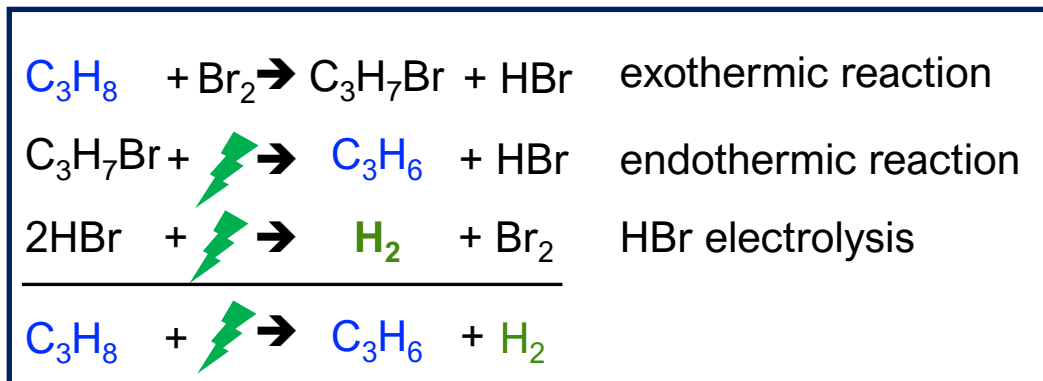
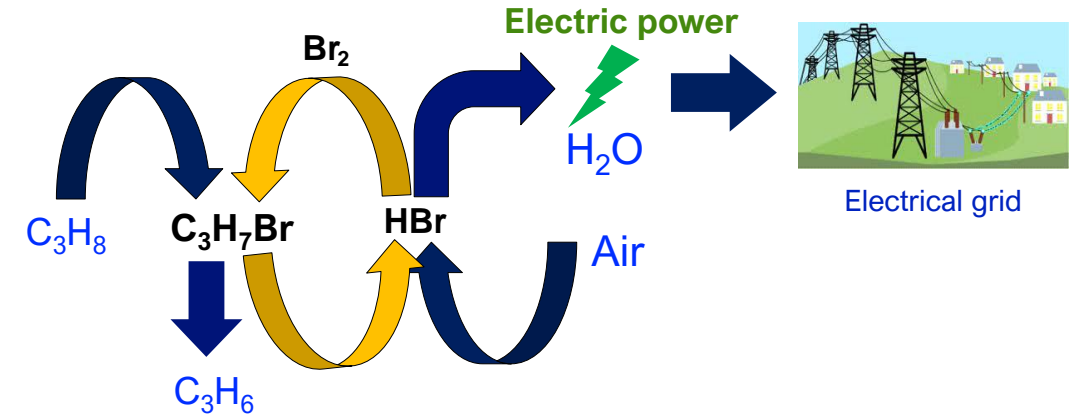


# Propane to Propylene technology

P2P with co-production of hydrogen

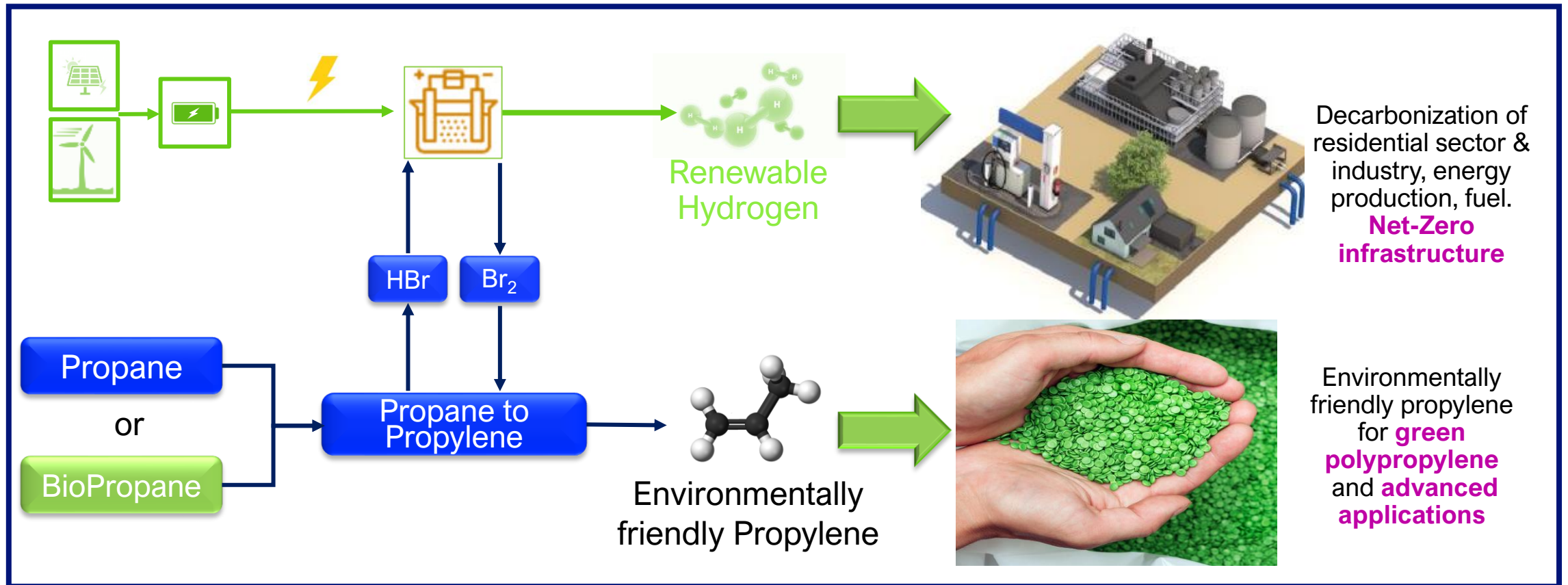


P2P with co-production of zero-emission energy





# Value proposition with Propane to Propylene technology

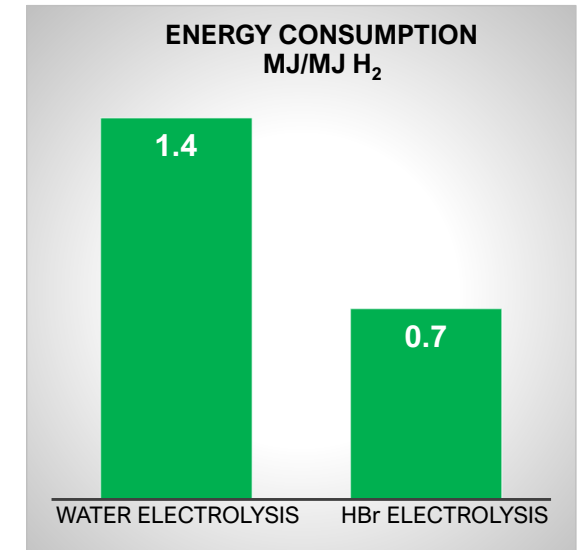
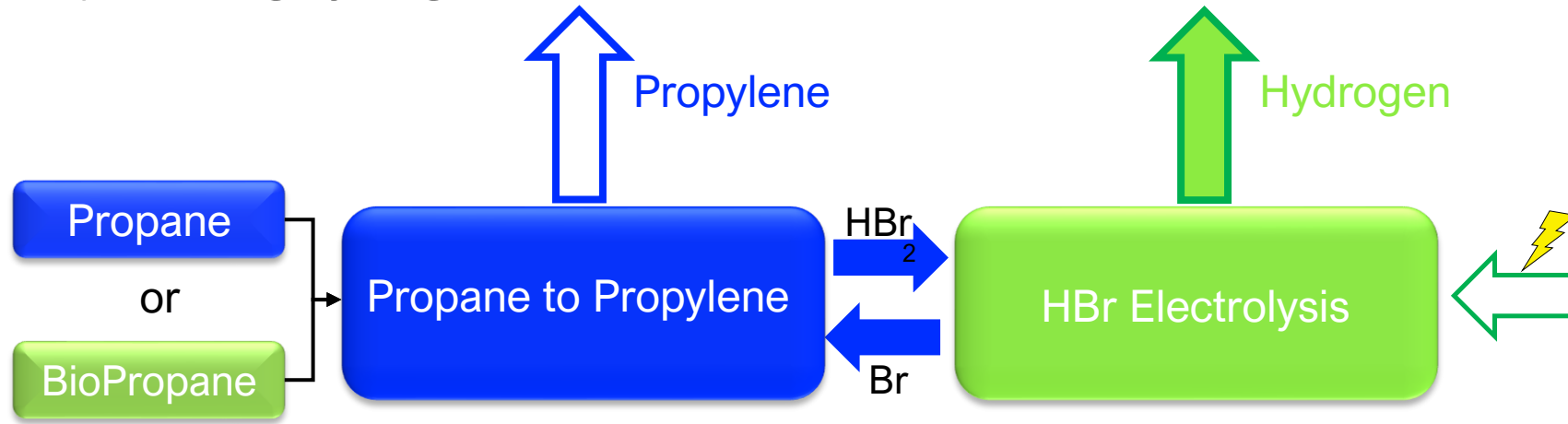


> P2P is a low-carbon process for producing propylene from propane that simultaneously generates renewable hydrogen, using half as much energy as water electrolysis



# Propane to Propylene in a nutshell

Low carbon, electrified technology for obtaining environmentally friendly propylene while co-producing hydrogen



- Eliminates propane – propylene equilibrium constraint
  - ➔ no C3-splitter required for polymer grade propylene, attractive for small to medium production scale
- 200°C lower reaction temperature than direct propane dehydrogenation
  - ➔ very high selectivity for propylene, complete conversion per pass
- Significantly lower GHG emissions than in direct propane dehydrogenation
- **Hydrogen** is produced electrochemically in an environment free of hydrocarbons
  - ➔ 100% electrified process and no need for downstream separation and purification

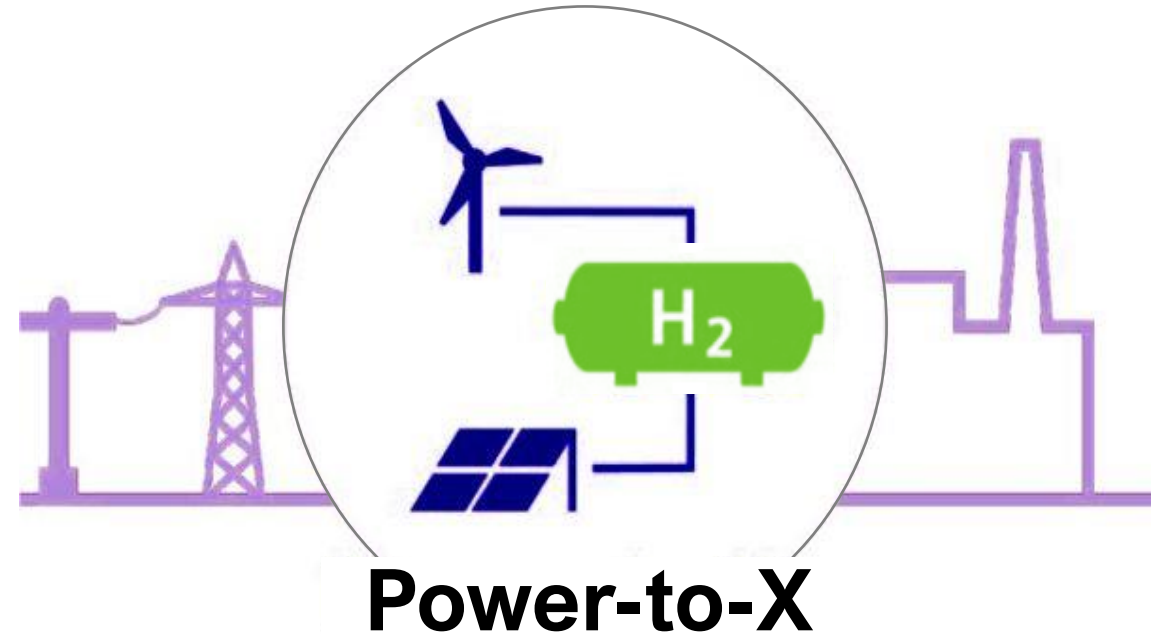
# Propane to Propylene technology benchmark

Technology	Propane to Propylene	Propane Dehydration – Technologies			
Licensors	Sulzer	A	B	C	D
Reaction temperature, °C	300-400	570-650	620-650	510-580	650
Catalyst type	non noble metal catalyst	CrO <sub>x</sub> /Al <sub>2</sub> O <sub>3</sub>	Sn-Pt-Cl-Al <sub>2</sub> O <sub>3</sub>	Pt-Sn-Zn/Ca aluminate	Pt-Ga-P/Al <sub>2</sub> O <sub>3</sub>
WHSV, h <sup>-1</sup>	8	1	-	6	-
Time between regenerations	>1 month	10 min	10 days	7 hours	2 min
Yield, wt% C-basis	>92	84-85	84-86	81-82	>90

Source: AIChE 2016, FCDh Paper

# Propane to Propylene - Part of Power-to-X

- Power-to-X (P2X) as a key technology in sector coupling, offers the possibility of converting energy from renewable resources into a chemical form and thus storing it for independent use, E-Hydrogen is the core component
- The X in the terminology refers to one of the following conversion targets: ammonia, chemicals, fuel, gas, hydrogen, liquids, methane, food, power or syngas
- Propane to Propylene is the second step in P2X, using renewable energy for production of green hydrogen and coproduction of high-quality carbon for advanced applications



<https://assignmentpoint.com/power-to-x/>

> Power-to-X refers to the use of sustainably produced electricity to generate storable energy, chemicals and other essential products while minimizing GHG emissions.

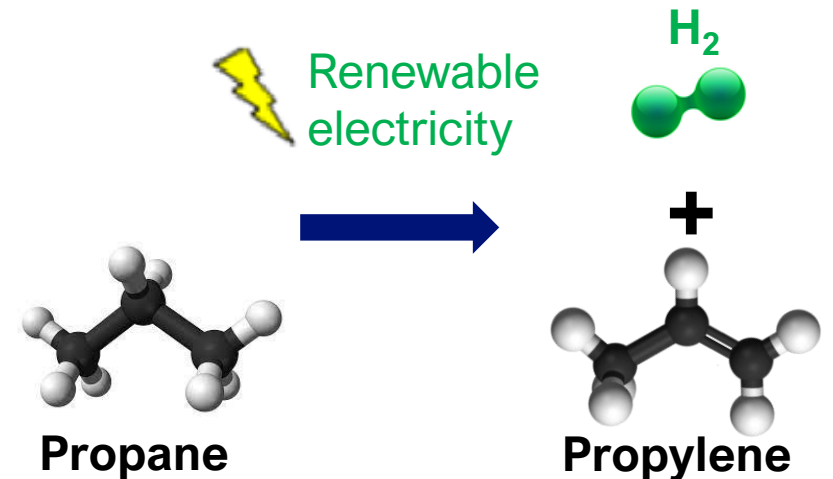
# Light alkanes for green hydrogen

- “Low-carbon H<sub>2</sub>” standard sets a “cradle-to-gate” emissions threshold of 36.4 gCO<sub>2</sub>/MJH<sub>2</sub>
- “Green H<sub>2</sub>” meets the criteria for low-carbon H<sub>2</sub> and, moreover, is produced exclusively using renewable energy



Source: EU Directive CertifHy-SD Hydrogen criteria

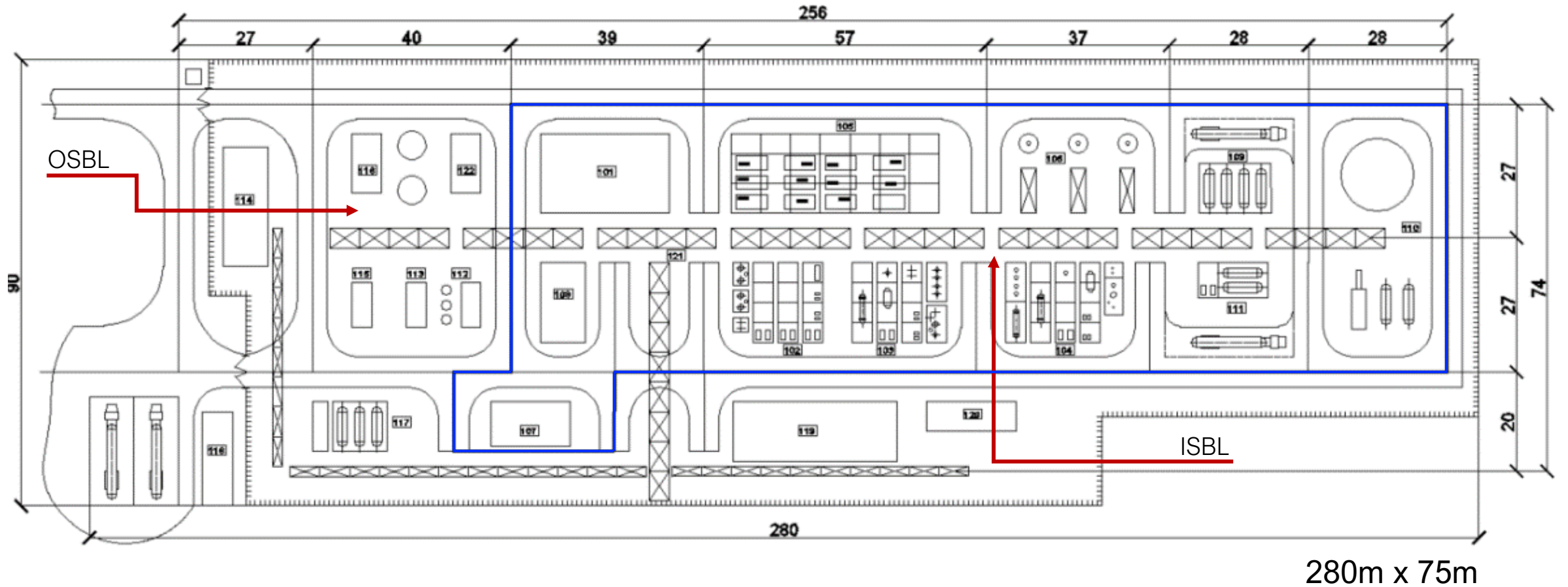
- It is not the source of feedstock for hydrogen that matters  
It is the environmental impact and the energy to obtain it
- Electrified processes to produce H<sub>2</sub> from propane become particularly attractive in near future due to credits for low carbon & green hydrogen production





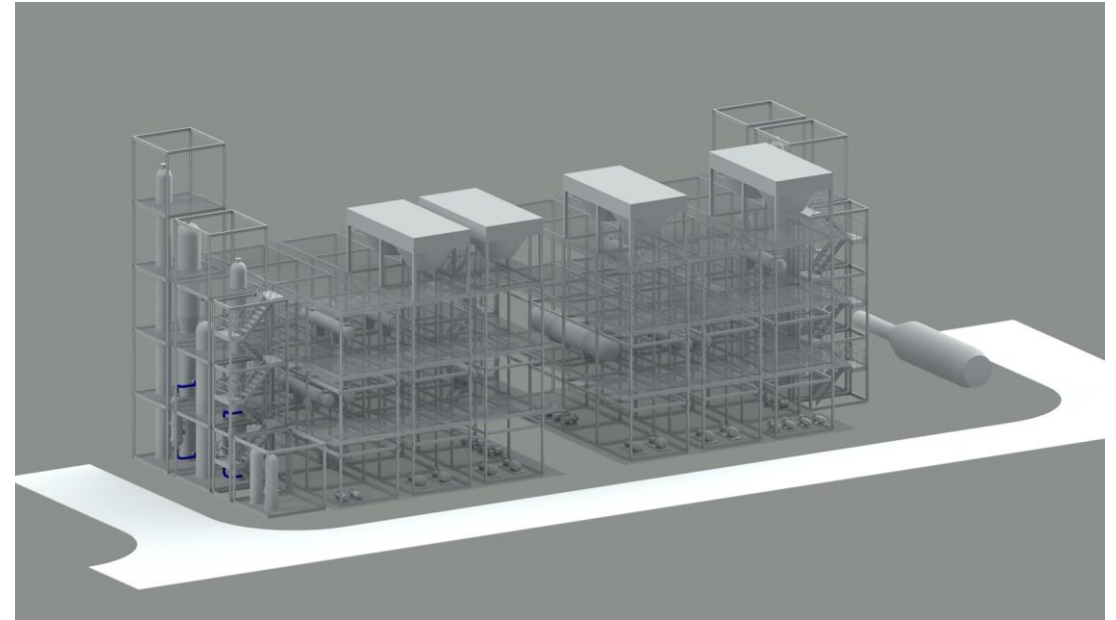
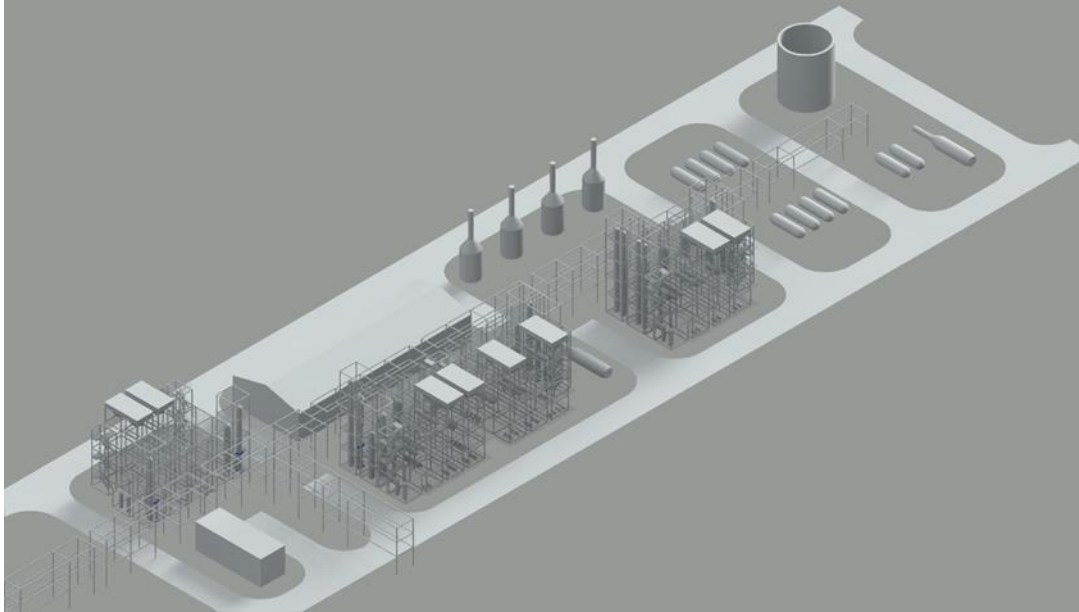
# Plant plot industrial scale

Typical





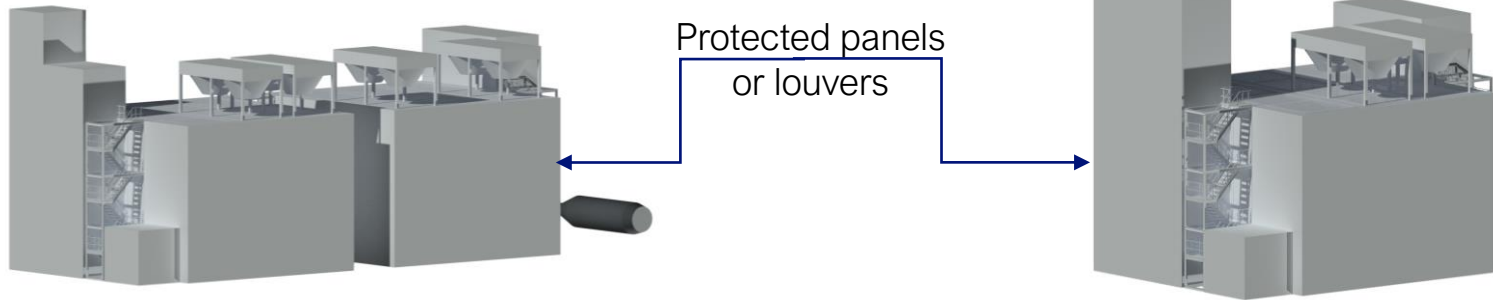
## 3-D Impressions



Plant installation will be adapted to local, project individual site conditions



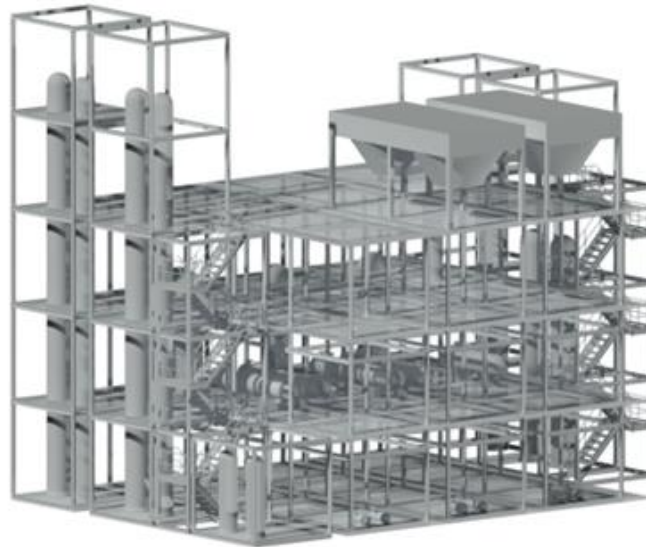
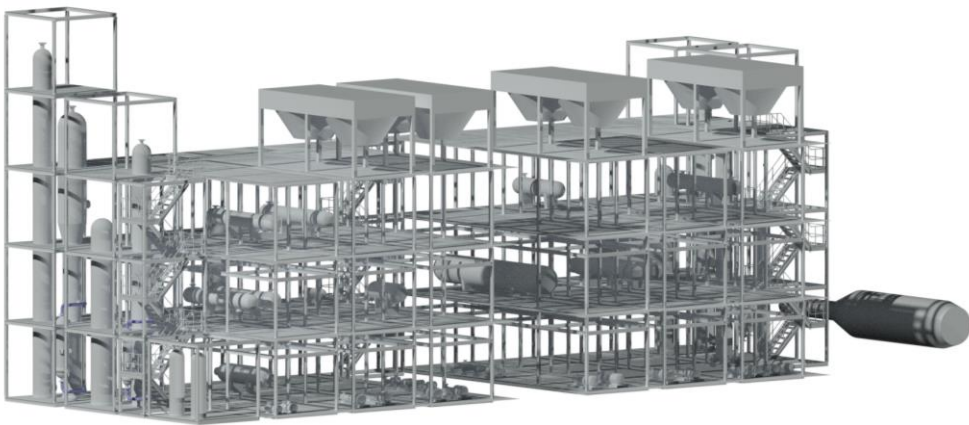
# Design details



Bromination section

Bromine recovery section

Synthesis section



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# Contact



Clean Fuels and Chemical Licensing



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